

On the Dissolution of Metals in Acids, by J. N. Brönsted and N. L. Ross Kane.

Page 3630. In line 2, and in line 7 of the second paragraph of fine print (occurring twice), the word *molal* is used when the word should be *molar*, in accordance with the usage on page 3629, line 4 of the second paragraph under 3, "The Decomposition of Sodium Amalgam." Calculation shows that the concentration is expressed in molar figures. I am adhering here to the Lewis system, in which molal refers to the number of moles per 1000 g. of solvent, and molar to the number of moles per 1000 cc. of solution. It is easily seen that in the case of mercury solutions the difference between the two is very great.

Pages 3642 and 3643. Figure 4 should be *phenol, aqueous buffer solution*, and Fig. 5 should be *benzene solution*.

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